# Physical Science Final Exam Study Guide

# Fall 2015

## 1) An Introduction to Physical Science

- a) An Introduction to The Scientific Method
  - i) Scientific Fact
    (1) A Definition of Fact
    (2) Facts Can Change
  - ii) The Scientific Method(1) A Definition of the Scientific Method
  - iii) Scientific Observation
    - (a) A Definition of Observation
    - (b) Two Classifications of Observations
      - (i) Qualitative Observations
      - (ii) Quantitative Observations
    - (C) Two Characteristics of Good Scientific Observation
      - (i) Good Scientific Observation Uses as Much Detail as Possible
      - (ii) Good Scientific Observation Uses as Many of the Five Senses as Possible
  - iv) Hypothesis
    - (1) A Definition of Hypothesis
    - (2) A Definition of *Scientific* Hypothesis
  - v) Scientific Law
    - (1) A Definition of Scientific Law
    - (2) Scientific Laws Can Change
  - vi) Scientific Theory

- (1) A Definition of Theory
- vii) Experimentation
  - (1) The Role of Experiment in Science
- viii) A Definition of Modern Science
- ix) A Definition of Pseudoscience
- x) Comparison/Contrast of Science and Philosophy (or Religion)

xi) The Scientific Revolution

- b) The SI System of Units
  - i) What is the SI System of Units?
  - ii) Who Uses the SI System?
  - iii) The SI System Versus the Metric System
  - iv) SI Prefixes
  - V) An Introduction to Unit Conversion
- c) Measurement and Error
  - i) Random Errors and Systematic Errors
  - ii) Accuracy Versus Precision
  - iii) Significant Figures
    - (1) Counting Significant Figures
    - (2) Using Significant Figures in Calculations

#### 2) An Introduction to Physics

- a) Position, Distance, and Displacement
  - i) Scalar Quantities
    - (1) A Definition of Scalar Quantity
    - (2) A List of Some Common Scalar Quantities
  - ii) Vector Quantities
    - (1) A Definition of Vector Quantity
    - (2) A List of Some Common Vector Quantities
  - iii) Position

- (1) A Definition of Position
- (2) An Abbreviation for Position
- (3) Position is a Vector Quantity
- (4) The SI Unit for Position
- (5) Position as Represented in Two Dimensions
- (6) Position as Represented in One Dimension
- iv) Distance
  - (1) A Definition of Distance
  - (2) An Abbreviation for Distance
  - (3) The SI Unit for Distance
  - (4) Distance is a Scalar Quantity
  - (5) Finding Distance Traveled
- v) Displacement
  - (1) A Definition of Displacement
  - (2) Finding Net Displacement Given Path
  - (3) An Abbreviation for Displacement
  - (4) The SI Unit for Displacement
  - (5) Displacement is a Vector Quantity
  - (6) A Formula for Displacement
  - (7) Finding Displacement Mathematically
- b) Speed and Velocity
  - i) Speed
    - (1) A Definition of Speed
    - (2) An Abbreviation for Speed
    - (3) A Formula for Average Speed
    - (4) The SI Units for Speed
    - (5) Speed is a Scalar Quantity
    - (6) Using Average Speed, Distance, and Time in Calculations
    - (7) Instantaneous Speed Versus Average Speed

- ii) Velocity
  - (1) A Definition of Velocity
  - (2) An Abbreviation for Velocity
  - (3) A Formula for Average Velocity
  - (4) The SI Units for Velocity
  - (5) Velocity is a Vector Quantity
  - (6) Using Average Velocity, Displacement, and Time in Calculations
  - (7) Instantaneous Velocity Versus Average Velocity
- c) Acceleration and Relative Motion
  - i) Constant Acceleration
    - (1) A Definition of Acceleration
    - (2) An Abbreviation for Acceleration
    - (3) A Formula for Average Acceleration
    - (4) The SI Units for Acceleration
    - (5) Acceleration is a Vector Quantity
    - (6) Using Average Acceleration, Change in Velocity, and Time in Calculations
    - (7) Galileo and Acceleration
    - (8) Free-Fall Acceleration
    - (9) Hang Time
  - ii) Relative Motion
- d) An Introduction to Force
  - i) Systems
    - (1) A Definition of a System
    - (2) Identifying Good Systems
  - ii) Net Force
    - (1) Finding Net Force in One Dimension
      - (a) Finding Net Force in One Dimension Mathematically

(b)Finding Net Force in One Dimension by Using Vector Addition/Subtraction

- (2) Finding Net Force in Two Dimensions
  - (a) Finding Net Force in Two Dimensions (in Component Form) Mathematically
  - (b) Finding Net Force in Two Dimensions by Using Vector Addition/Subtraction
- iii) Mechanical Equilibrium
  - (1) Net Force on an Object in Mechanical Equilibrium is Equal to Zero (for an Non-Rotating System)
  - (2) Two Types of Mechanical Equilibrium
    (a) Static Equilibrium
    (b) Dynamic Equilibrium
- iv) Some Types of Forces
  - (1) Weight
  - (2) Support Force
  - (3) Frictional Force
- e) Newton's Three Laws of Motion
  - i) Newton's First Law of Motion
    - (1) Inertia
    - (2) Mass
    - (3) A Description of Newton's First Law of Motion
  - ii) Newton's Second Law of Motion
    - (1) A Description of Newton's Second Law of Motion
    - (2) Free-Fall Acceleration and Newton's Second Law of Motion
    - (3) Non-Free-Fall Motion and Newton's Second Law of Motion
      - (a) Terminal Velocity/Speed

- iii) Newton's Third Law of Motion
  - (1) Interaction
    - (a) Action/Reaction Pairs of Forces
  - (2) A Description of Newton's Third Law of Motion
- iv) Newton's Second and Third Laws Involving Different Size Masses
- f) Momentum and Impulse
  - i) Momentum
    - (1) Momentum is "Inertia in Motion"
    - (2) An Abbreviation for Momentum
    - (3) A Formula for Momentum
    - (4) The SI Units for Momentum
    - (5) Momentum is a Vector Quantity
    - (6) Using Momentum, Mass, and Velocity in Calculations
  - ii) Impulse
    - (1) Impulse is "Change in Momentum"
    - (2) An Abbreviation for Impulse
    - (3) A Formula for Impulse
    - (4) The SI Units for Impulse
    - (5) Impulse is a Vector Quantity
    - (6) Using Impulse (or Change in Momentum), Force, and Time in Calculations
    - (7) Examples of Impulse Changing Momentum
      (a) Impulse Increasing Momentum
      (b) Impulse Decreasing Momentum
- g) The Law of Conservation of Linear Momentum
  - i) English Statement of the Law of Conservation of Linear Momentum
  - ii) A Definition of a Closed System

(1) Linear Momentum is Conserved Only for a Closed System

- iii) Mathematical Statement of the Law of Conservation of Linear Momentum
- iv) Using the Law of Conservation of Linear Momentum to
   Find Momentum, Mass, and/or Velocity of Objects in a
   Closed System
- h) Energy and Work
  - i) Energy
    - (1) A Definition of Energy
    - (2) An Abbreviation for Energy
    - (3) Some Common Forms of Energy
      - (a) Kinetic Energy
        - (i) A Definition of Kinetic Energy
        - (ii) A Formula for Kinetic Energy
      - (b)Potential Energy
        - (i) A Definition of Potential Energy
        - (ii) Some Common Forms of Potential Energy
          - 1. Elastic Potential Energy
          - 2. Gravitational Potential Energy
            - a. A Formula for Gravitational Potential Energy
    - (4) The SI Unit for Energy
    - (5) Energy is a Scalar Quantity
    - (6) Using Energy and Other Quantities in Calculations
  - ii) Work
    - (1) Work is "Change in Energy"
    - (2) An Abbreviation of Work
    - (3) A Formula for Work
    - (4) The SI Unit for Work
    - (5) Work is a Scalar Quantity

- (6) Using Work (Change in Energy), Force, and Distance in Calculations
- i) The Law of Conservation of Energy
  - i) English Statement of the Law of Conservation of Energy
     (1) Energy for a System is Conserved Only when Net Work
     Done on or by the System is Equal to Zero
  - ii) Using the Law of Conservation of Energy in Calculations
- j) Power, Efficiency, and Machines
  - i) Power
    - (1) A Definition of Power
    - (2) An Abbreviation for Power
    - (3) A Formula for Power
    - (4) The SI Units for Power
    - (5) Power is a Scalar Quantity
    - (6) Using Power, Work (Change in Energy), and Time in Calculations
  - ii) Efficiency
    - (1) A Definition of Efficiency
    - (2) An Abbreviation for Efficiency
    - (3) A Formula for Efficiency
    - (4) Efficiency is Dimensionless (It Has No Units)
    - (5) Efficiency is a Scalar Quantity
    - (6) Using Efficiency, Work Input (or Energy Input), and Work Output (or Energy Output) in Calculations
  - iii) Machines
    - (1) A Definition of Machine
    - (2) Machines Follow the Law of Conservation of Energy
- k) Gravity, Projectiles, and Satellites
  - i) Gravity

(1) Newton's Law of Universal Gravitation

- (a) Mathematical Statement of Newton's Law of Universal Gravitation
- (b) The Inverse Square Law
- (C) How Changing Mass and/or Distance Affect Gravitational Force

(2) Weight

- (a) A Definition of Weight
- (3) Apparent Weight
  - (a) A Definition of Apparent Weight
  - (b) Some Cases in Which Apparent Weight is not Equal to Actual Weight
    - (i) Apparent Weight is Greater than Actual Weight for an Object Which is Accelerating Upward
    - (ii) Apparent Weight is Less than Actual Weight for an Object Which is Accelerating Downward
    - (iii) An object is Weightless When it has No
       Support Force (When it is Accelerating Downward at g)
- ii) Projectile Motion
  - (a) A Definition of a Projectile
  - (b) The Velocity of a Projectile has Two Components
    - Vertical Component of Velocity Undergoes an Acceleration of g Downward (for Negligible Air Resistance)
    - (ii) Horizontal Component of Velocity is Constant (for Negligible Air Resistance)

iii) Satellite Motion

(1) Energy is Conserved for Satellite Motion

(2) Escape Speed

- 1) Temperature and Thermal Energy
  - i) Temperature
    - (1) A Definition of Temperature
    - (2) An Abbreviation for Temperature
    - (3) The SI Unit for Temperature
    - (4) Temperature is a Scalar Quantity
    - (5) Absolute Zero
  - ii) Thermal Energy
    - (1) A Definition of Thermal Energy
    - (2) An Abbreviation for Thermal Energy
    - (3) Some Acceptable Units for Thermal Energy
    - (4) Thermal Energy is a Scalar Quantity
  - iii) Heat
    - (1) A Definition of Heat
    - (2) An Abbreviation for Heat
    - (3) Some Acceptable Units for Heat
    - (4) Heat is a Scalar Quantity
  - iv) Specific Heat Capacity
    - (1) A Definition of Specific Heat Capacity
    - (2) An Abbreviation for Specific Heat Capacity
    - (3) The SI Units for Specific Heat Capacity
    - (4) Specific Heat Capacity is a Scalar Quantity
    - (5) Using Specific Heat Capacity, Thermal Energy, Mass, and Change in Temperature in Calculations
  - v) Thermal Expansion
    - (1) A Definition of Thermal Expansion
    - (2) Typical Behavior of Substances Undergoing Changes in Temperature

(3) The Thermal Expansion of Water

- vi) Phase Changes
  - (1) The Five Known Phases of Matter
    - (a) Bose-Einstein Condensate
    - (b) Solid
    - (C) Liquid
    - (d) Gas
    - (e) Plasma
  - (2) Common Types of Phase Changes
    - (a) Melting and Freezing
    - (b) Evaporation and Condensation
    - (C) Sublimation and Deposition
- vii) Energy and Phase Changes
- m) Thermal Energy Transfer
  - i) The Three Processes of Heat Transfer
    - (1) Conduction
      - (a) A Definition of Conduction
      - (b) Conduction can Occur in Matter in Certain Phases
      - (C) Good Conductors
      - (d) Poor Conductors a.k.a. Insulators
    - (2) Convection
      - (a) Definition of Convection
      - (b) Convection can Occur in Matter in Certain Phases
    - (3) Radiation
      - (a) A Definition of Radiation
      - (b) Radiation can Occur in Any Phase of Matter
      - (C) The Electromagnetic Spectrum
  - ii) The Greenhouse Effect
  - iii) Reducing Heat Transfer in a Thermos

- n) Static Electricity
  - i) Electric Charge
  - ii) Coulomb's Law
  - iii) Electric Field
  - iv) Electric Potential
  - v) Voltage Sources
- 0) Current Electricity
  - i) Electric Current
  - ii) Electrical Resistance
  - iii) Ohm's Law
  - iv) Electric Circuits
  - V) Electric Power
- p) Magnetism
  - i) Magnetic Poles
  - ii) Magnetic Fields
  - iii) Magnetic Domains
  - iv) Electric Currents and Magnetic Fields
  - v) Magnetic Forces on Moving Charges
  - vi) Electromagnetic Induction
  - vii) Generators and Alternating Current
  - viii) Power Production
  - ix) The Transformer—Boosting or Lowering Voltage x) Field Induction

#### 3) An Introduction to Chemistry

- a) An Introduction to Chemical Concepts
  - i) Şizes
    - (1) Submicroscopic
    - (2) Microscopic
    - (3) Macroscopic

- ii) Physical and Chemical Properties
- iii) Physical and Chemical Changes
- iv) Elements Versus Compounds
- v) Naming Compounds
  (1) Naming Type I Ionic Compounds
  (2) Naming Covalent Compounds
- b) Atoms
  - i) Parts of the Atom
    - (1) The Electron
    - (2) The Atomic Nucleus
      - (a) Protons
      - (b) Neutrons
  - ii) Bohr's Model of the Atom
- c) The Periodic Table of Elements
  - i) Organizing the Elements
    - (a) Metals, Nonmetals, and Metalloids
    - (b) Atomic Groups
    - (C) Atomic Periods
- d) Chemical Bonding
  - i) Electron-Dot Structures
  - ii) Ions
  - iii) Types of Chemical Bonds
    - (1) Jonic Bonds
    - (2) Covalent Bonds
  - iv) Chemical Equations
    - (1) The Mole
    - (2) Balancing Simple Chemical Equations

## 4) Weather

a) An Introduction to Weather

- i) Atmospheric Moisture
- ii) Weather Variables
- iii) Cloud Development
- iv) Air Masses, Fronts, and Storms
- v) Violent Weather
- vi) Weather ForeCasting

#### 5) A Brief Introduction to Astronomy

- a) The Solar System
  - i) How Did the Solar System Form?
  - ii) Objects in the Solar System
    - (1) The Sun
    - (2) The Planets and Moons
      - (a) The Inner Planets
        - (i) Earth's Moon
      - (b) The Outer Planets
    - (3) Other Objects
      - (a) Comets
      - (b) Asteroids
      - (C) Meteoroids, Meteors, and Meteorites
- b) Stars and Galaxies
  - i) Stars
    - (1) The Life of a Star
      - (a) The Hertzsprung-Russell Diagram
      - (b) The Birth of a Star
      - (C) The Life of a Star
      - (d) The Death of a Star
  - ii) Galaxies